

(2E,4E)-1-(3-Nitrophenyl)-5-phenylpenta-2,4-dien-1-one**P. S. Patil,^a Jeannie Bee-Jan Teh,^b
Hoong-Kun Fun,^{b*}
Ibrahim Abdul Razak^b and
S. M. Dharmaprabakash^a**^aDepartment of Studies in Physics, Mangalore University, Mangalagangothri, Mangalore 574 199, India, and ^bX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia

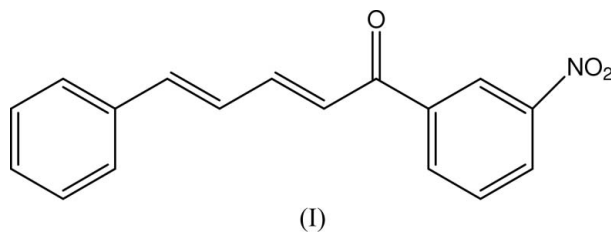
Correspondence e-mail: hkfun@usm.my

Key indicatorsSingle-crystal X-ray study
 $T = 100\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$
 R factor = 0.042
 wR factor = 0.115
Data-to-parameter ratio = 18.5For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound, $\text{C}_{17}\text{H}_{13}\text{NO}_3$, the dihedral angle between the benzene rings is $2.57(4)^\circ$. The crystal structure is stabilized by $\text{C}-\text{H}\cdots\text{O}$ interactions, which form a three-dimensional network, and by $\text{C}-\text{H}\cdots\pi$ interactions.

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Lasers are used increasingly for high-speed recording, storage, information transmission and processing. Many electro-optical devices rely on the use of non-linear optical (NLO) materials, in particular those with the capability of frequency conversion (second-harmonic generation, SHG) (Chemla & Zyss, 1987). Among several organic compounds reported for their NLO properties, chalcone derivatives are prominent materials due to their excellent blue-light transmittance and ease of forming crystals. They provide the configuration necessary to show NLO activity, with two planar rings connected through a conjugated double bond (Patil, Teh *et al.*, 2006; Patil, Ng *et al.*, 2007; Patil, Rosli *et al.*, 2007). The title compound, (I), was prepared as part of our ongoing studies on non-linear optical materials (Patil, Dharmaprabakash *et al.*, 2006; Shettigar *et al.*, 2006; Chantrapromma, Jindawong *et al.*, 2006; Chantrapromma, Ruanwas *et al.*, 2006). Crystals of the title compound, (I) (Fig. 1), should exhibit second-order NLO properties as they crystallize in a non-centrosymmetric space group.



Bond lengths and angles in (I) have normal values (Allen *et al.*, 1987). The least-squares plane through the C7–C11/O1

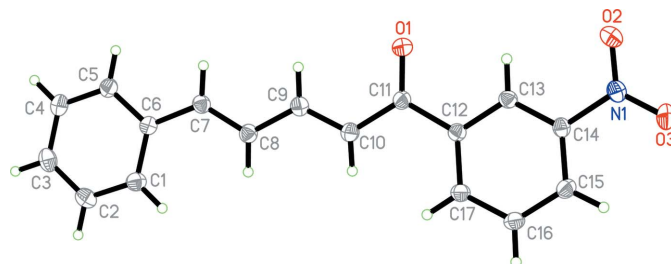


Figure 1
The molecular structure of (I), showing 50% probability displacement ellipsoids and the atom numbering.

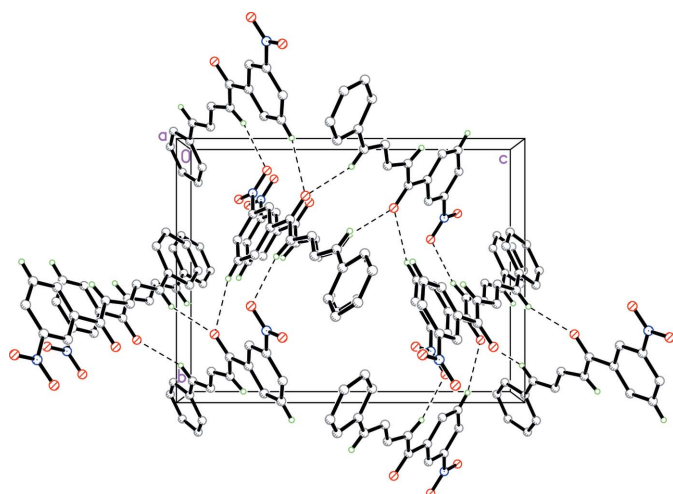


Figure 2

The crystal packing of (I), viewed down the *a* axis. Dashed lines indicate hydrogen bonds. H atoms not involved in hydrogen bonding have been omitted for clarity.

group makes dihedral angles of 7.04 (4) and 8.70 (4)° with the C1–C6 and C12–C17 benzene rings, respectively. The dihedral angle between these rings is 2.57 (4)° and the nitro group at C14 lies close to the C12–C17 ring plane, with O2–N1–C14–C13 and O3–N1–C14–C15 torsion angles of 0.17 (18) and 1.60 (19)°, respectively.

The crystal structure (Fig. 2) is stabilized by C–H···O interactions (Table 1), which form a three-dimensional network, and is further consolidated by C–H··· π interactions (Table 1) involving the C1–C6 benzene ring.

Experimental

Cinnamaldehyde (0.01 mol) and 3-nitroacetophenone (0.01 mol) were stirred in methanol (60 ml) at room temperature. 10% aqueous NaOH solution (5 g) was added and the mixture stirred for 2 h. The resulting precipitate was filtered, washed with water, dried and the crude product recrystallized from acetone. Crystals suitable for X-ray analysis were grown by slow evaporation of an acetone solution at room temperature.

Crystal data

C ₁₇ H ₁₃ NO ₃	$V = 1378.38$ (4) Å ³
$M_r = 279.28$	$Z = 4$
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation
$a = 7.0167$ (1) Å	$\mu = 0.09$ mm ⁻¹
$b = 12.1953$ (2) Å	$T = 100.0$ (1) K
$c = 16.1081$ (3) Å	$0.47 \times 0.26 \times 0.14$ mm

Data collection

Bruker SMART APEX2 CCD diffractometer	38028 measured reflections
Absorption correction: multi-scan (SADABS; Bruker, 2005)	3523 independent reflections
$T_{\min} = 0.864$, $T_{\max} = 0.988$	3097 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.044$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.042$	190 parameters
$wR(F^2) = 0.115$	H-atom parameters constrained
$S = 1.07$	$\Delta\rho_{\text{max}} = 0.29$ e Å ⁻³
3523 reflections	$\Delta\rho_{\text{min}} = -0.21$ e Å ⁻³

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C7–H7A···O1 ⁱ	0.93	2.42	3.297 (2)	157
C10–H10A···O2 ⁱⁱ	0.93	2.47	3.358 (2)	159
C16–H16A···O1 ⁱⁱ	0.93	2.57	3.462 (2)	161
C15–H15A···Cg1 ⁱⁱⁱ	0.93	2.76	3.612 (1)	152

Symmetry codes: (i) $x - \frac{1}{2}, -y + \frac{3}{2}, -z$; (ii) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$; (iii) $-x + \frac{1}{2}, -y + 2, z + \frac{1}{2}$. Cg1 is the centroid of the C1–C6 ring.

In the absence of significant anomalous scattering effects, 2727 Friedel pairs were merged. H atoms were positioned geometrically and treated as riding, with C–H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: APEX2 (Bruker, 2005); cell refinement: APEX2; data reduction: SAINT (Bruker, 2005); program(s) used to solve structure: SHELXTL (Sheldrick, 1998); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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